

Development of Lipid-Based Nano Emulsions of Poorly Soluble Drug Nebivolol

Praveen.G*, Nagulu.M**

*Research Scholar, Mewar University, Chittorgarh-312901, Rajasthan, India

*E-mail: pavvi2pavvi@gmail.com

**Research Supervisor, Mewar University, Chittorgarh-312901, Rajasthan, India

*Corresponding Author

G. Praveen

Mobile No: +91-6302241681

E-mail: pharmaresearch77@gmail.com

Abstract

The main purpose of this study was to investigate the potential of self-nano emulsified drug delivery system (SNEDDS) to improve the solubility and dissolution of nebivolol. SNEDDS was developed by using rational blends of excipients with good solubilizing ability for nebivolol which was selected based on solubility studies. Further phase diagrams were constructed to determine the self-emulsifying region. The SNEDDS formulations were evaluated for various tests and characterised for FTIR, DSC, SEM and stability studies. The optimal formulation (F14) with the best self-nano emulsified and solubilization ability consisted of 25% (w/w) corn oil as oil, 57% (w/w) caproic acid as surfactant and 18% (w/w) PEG600 as cosurfactant. The formulation displayed maximum drug content of 99.23%, 98.51% entrapment efficiency and drug release of 99.96% in 60 minutes. The average droplet size and zeta-potential of the optimal SNEDDS were 165.2 nm and -13.2 mv, respectively. The forced degradation study results show that the pH dependant degradation and the formulation F14 found to be stable after storage at accelerated conditions at $40\pm 2^\circ\text{C}/75\pm 5\%\text{RH}$ for a period of six months. Hence SNEDDS may be considered as suitable formulation approach for improving the solubility, dissolution and therapeutic efficiency of nebivolol.

Keywords: Nebivolol, Hypertension, Self-nanoemulsified drug delivery system, Pseudoternary phase diagram, in-vitro drug release, Forced degradation studies

I. INTRODUCTION

Nanotechnology has become a buzzword for scientific experts, and efforts are ongoing to extend its applications in various medical and pharmaceutical aspects. The nanoscale technologies can be generally categorized into lipid-based nanocarriers, polymeric nanocarriers, inorganic nanocarriers, and drug nanoparticles or nanosuspensions. Within the lipid-based nanocarriers category, there has been a resurgence of interest in nanoemulsions since low-energy emulsification methods, such as spontaneous or self-nanoemulsification, have been developed. Self-nanoemulsifying drug delivery systems (SNEDDS) are anhydrous homogeneous liquid mixtures, composed of oil, surfactant, drug, and/or cosolvents, which spontaneously form transparent nanoemulsion

(20–200 nm droplet size) upon aqueous dilution with gentle agitation [1], [2].

A vital feature of a successful SNEDDS formulation is its capability to hold the drug in solution, throughout the GIT, for sufficient time to allow for absorption. Many poorly water-soluble drugs have high solubility in SNEDDS formulations but could make a risk of precipitation after aqueous dispersion of the formulation or during its digestion in the intestine [3]

The model drug for the current study had been selected from the biopharmaceutical classification system class II. Nebivolol hydrochloride is chemically known as α , α - [iminobis (methylene)] bis [6-fluoro-3,4-dihydro-2H-1-benzopyran-2-methanol] hydrochloride. It is a highly selective β 1-blocker with nitric oxide mediated vasodilatory actions and beneficial effects on vascular endothelial function. Nebivolol is used in the management of hypertension. It is given by mouth as the hydrochloride although doses are expressed in terms of base. The usual dose is 10 mg and 5 mg daily. An initial dose of 2.5 mg daily is employed in the elderly and in patients with renal impairment [4]. The main objective of this work was to investigate the possibility of improving the solubility and dissolution rate of Nebivolol by developing suitable SNEDDS.

II. MATERIAL AND METHODS

A. *Materials used*

Nebivolol drug was purchased from Hetero drugs Ltd, Hyderabad. Sesame oil, olive oil, sunflower oil, soybean oil, palm kernel oil, palm oil, corn oil, peanut oil, peceol oil, acrysol EL135, acconon E, acconon Sorb20, cremophor CO 60, tween 20, tween 80, caprol PGE 860, lauroglycol 90, cremophor RH 40, labrasol, labrafac, caproic acid, cremophor EL, caproic acid, transcutool p, propylene glycol, capmul MCMC8, PEG 400 and PEG 600 were purchased from Gattefosse, Mumbai. All the reagents used were of analytical grade.

B. *Methods*

1) *Solubility of nebivolol in Vehicles*

Naturally occurring different vegetable oils, various surfactants and co-surfactants were studied for nebivolol solubility in order to identify the components for construction of ternary phase diagrams procedure as reported elsewhere. Supernatant was analyzed for nebivolol using UV spectrophotometer at 280nm. [5]

2) *Construction of Pseudo-Ternary Phase Diagrams*

From the solubility study oil, surfactant and co-surfactant were chosen based on the maximum solubility of the drug in it, the chosen vehicles were mixed in various ratios ranging from 1:9 to 9:1 (oil: Smix). Smix is the mixture of surfactant and co-surfactant prepared in defined ratios of 1:1, 2:1, and 3:1. The ratios with no phase separation and clear appearance with no turbidity were separated and checked for the transmittance using UV spectrophotometer [6]. The transmittance value more than 90 indicated nano size droplets formation hence these ratios were noted and used for plotting pseudo-ternary phase diagram [7] Pseudo ternary phase diagram is constructed using CHEMIX software.

3) *Effect of nebivolol loading*

In the view of this, effect of nebivolol loading on the transmittance, phase behaviour and area of nanoemulsion formation was studied on Corn oil - Caproic acid - PEG600 compositions with Smix in 3:1 ratio, which gave more area of nanoemulsification region among the other ratios.

Fifteen compositions of varying ratios of Corn oil - Caproic acid - PEG600 were taken and in 1ml composition of each ratio were incorporated with 2.5 mg, 5 mg and 10 mg of nebivolol (i.e., $15 \times 3 = 45$ formulations). [8]

4) *Preparation and Evaluation of Nebivolol SNEDDS*

A series of SNEDDS (F1- F14, the composition was shown in Table 1) which showed transmittance values more than 90) were selected from 5 mg loaded nebivolol system and prepared as described above [7]. About 1ml of the formulation (equivalent to 5 mg of the nebivolol) was filled in size '00' hard gelatin capsules, sealed and stored at ambient temperature (25° C) until used. These SNEDDS were evaluated for visual observations, turbidity, and robustness to dilution and in vitro dissolution study and were optimized [9]

TABLE 1: COMPOSITION OF NEBIVOLOL SNEDDS

S. No	Formulation code	Nebivolol drug (mg)	Ratios of Oil: Smix	Oil (Corn oil)	Smix 3:1	
					Surfactant (Caproic acid)	Co-surfactant (PEG 600)
1	F1	10	01:01	50	37.5	12.5
2	F2	10	01:02	33	49.5	16.5
3	F3	10	02:01	66	24.75	8.25
4	F4	10	03:01	75	18.75	6.25
5	F5	10	02:03	40	45	15
6	F6	10	02:05	28.5	53.25	17.75
7	F7	10	03:02	60	30	10
8	F8	10	03:04	42.6	42.6	14.8
9	F9	10	03:07	30	52.5	17.5
10	F10	10	04:03	57.1	31.95	10.65
11	F11	10	05:02	71	21.3	7.1
12	F12	10	05:03	62.5	28.12	9.3
13	F13	10	07:03	70	22.5	7.5
14	F14	10	01:03	25	56.25	18.75

C. Evaluations

Visual Observations, Turbidity Measurement, Robustness to Dilution, Percentage drug content, Entrapment efficiency. [10], [11].

D. In Vitro Dissolution Study

In vitro dissolution studies were conducted for nebivolol pure drug and nebivolol SNEDDS formulations (F1-F14) was performed using USP dissolution Apparatus II (Lab India DS 8000, Mumbai, India). Hard gelatin capsules, size "1" filled with nebivolol SNEDDS formulation were introduced into 900 mL of freshly prepared pH 6.8 phosphate buffer maintained at $37 \pm 0.5^\circ \text{C}$ and the speed of the paddle was set at 50 rpm. Capsules were held to the bottom of the vessel using copper sinkers. At pre-determined time intervals, 5 mL of samples were withdrawn by means of a syringe and immediately replaced with 5 mL of fresh medium maintained at $37 \pm 0.5^\circ \text{C}$. The samples were suitably diluted and analyzed for nebivolol using UV method spectrophotometrically at 280nm [13], [13]. For comparison, similarly dissolution studies of pure drug were also performed [14]. All measurements were done in triplicate [15].

E. Characterization of optimised nebivolol SNEDDS formulation

FTIR studies, Globule Size and Zeta potential [7], Scanning electron microscopy [16]

F. Forced Degradation and Accelerated Stability Studies

Nebivolol is very susceptible to decomposition when it is exposed to oxygen, light, temperature, humidity, carbon dioxide and acidic pharmaceutical excipients. The percentage degradation of nebivolol was calculated (ICH Harmonized Tripartite guideline on "Stability Testing of New Drug Substances and Products Q1A (R2)", 6 February 2003). [17]

G. Accelerated Stability Studies

All formulations filled in hard gelatin capsules were packed in HDPE screw capped bottles and kept in humidity chambers maintained at $40 \pm 2^\circ\text{C} / 75 \pm 5\% \text{RH}$ as per ICH guidelines for Zone III and stored for 6 months.

III. RESULTS AND DISCUSSION

H. Determination of nebigivol solubility in various excipients

Corn oil was selected as oil phase due to its higher solubilization ($3.05 \pm 0.69 \text{mg/g}$) of nebigivol compared to other oils (Figure 1).

Surfactant Caproic acid and co-surfactant PEG600 was selected for further studies due to their higher solubilizing capacity towards nebigivol [18], [19], [20] (Figure 2 and 3).

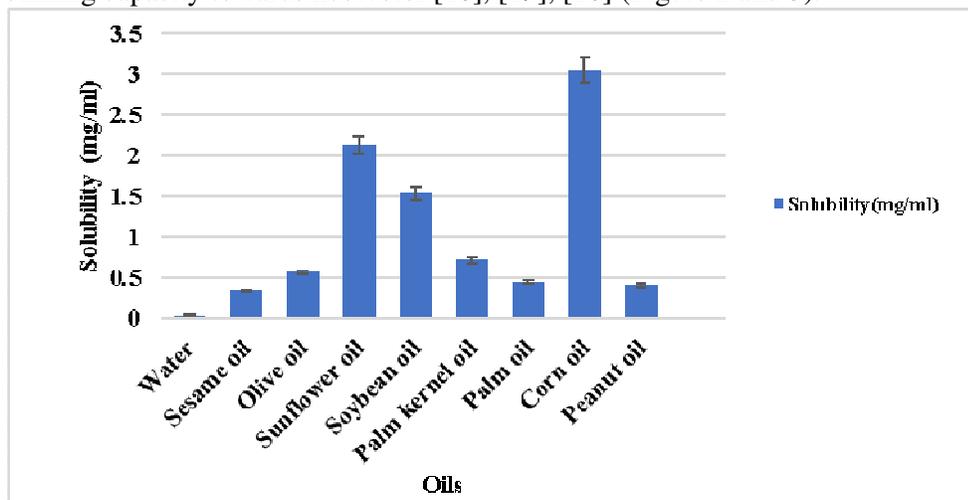


Fig 1: solubility of nebigivol in various Oils

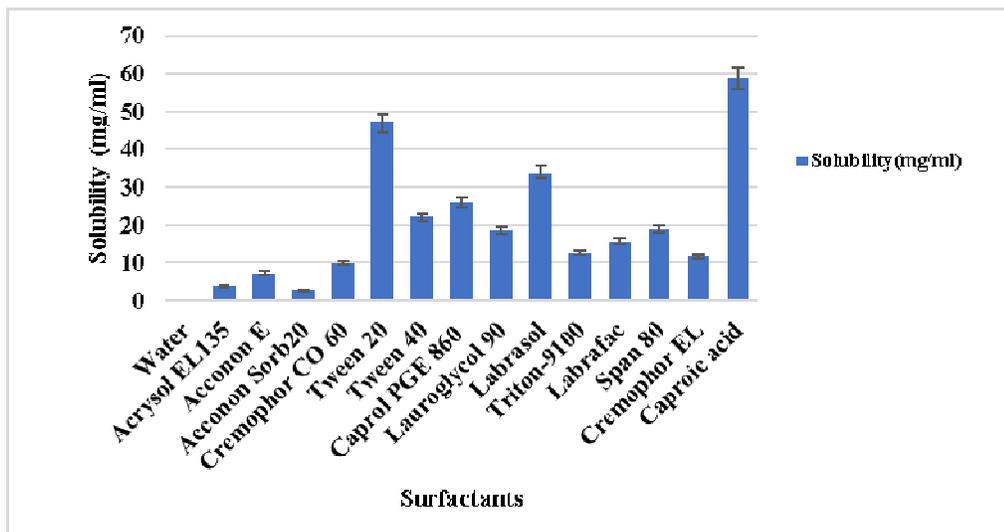


Fig 2: solubility of nebigivol in various Surfactants

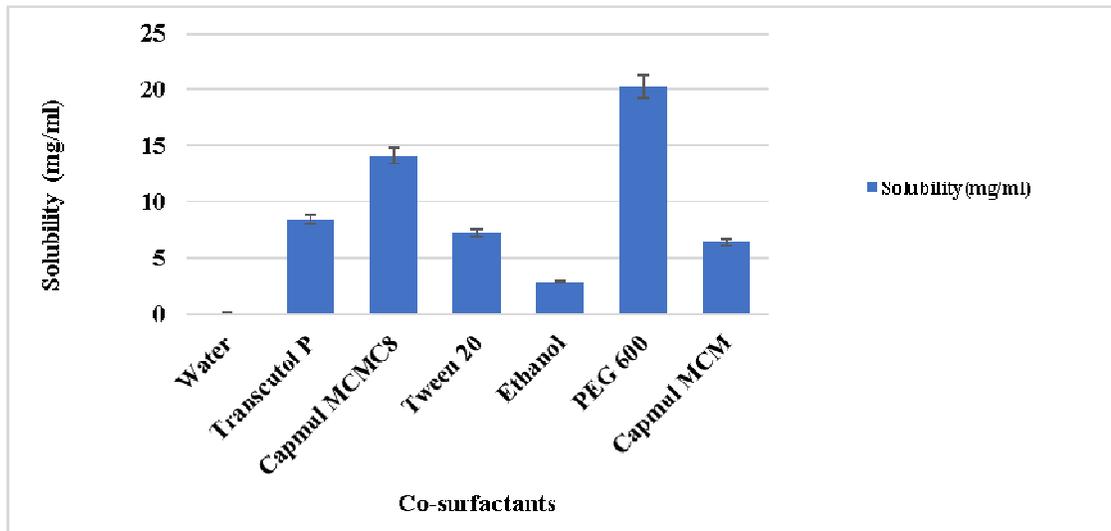


Fig 3: Solubility of nebigivolol in various Co-Surfactants

I. Construction of Ternary Phase Diagrams

The region of nano emulsification was indicated as shadow area encircled by a solid line and the points indicate the compositions of the system explored. Corn oil - Caproic acid - PEG600 system with Smix ratio in 3:1 exhibited larger nanoemulsification region (Fig. 6) as compared to 1:1 and 2:1 Smix ratio (Fig. 4 and 5).

CHEMIX School - TERNARY PLOT

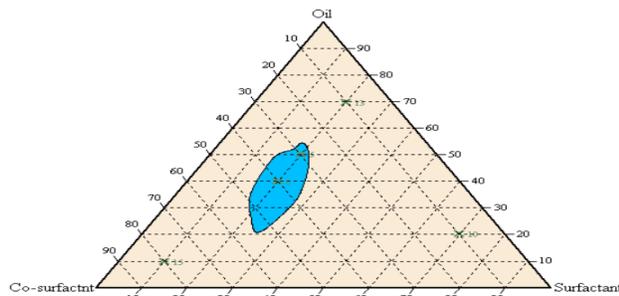


Fig. (4). Ternary phase diagram for Corn oil - Caproic acid - PEG600 with Smix in 1:1 ratio (Key: the filled region within the ternary phase diagram indicates nanoemulsification area where the transmittance is greater than 90)

CHEMIX School - TERNARY PLOT

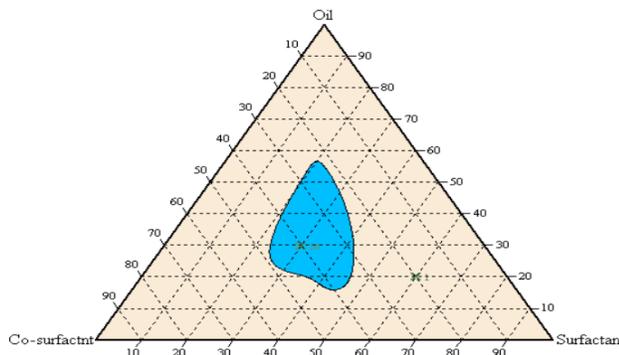


Fig. (5). Ternary phase diagram for Corn oil - Caproic acid - PEG600 with Smix in 2:1 ratio (Key: the filled region within the ternary phase diagram indicates nanoemulsification area where the transmittance is greater than 90).

CHEMIX School - TERNARY PLOT

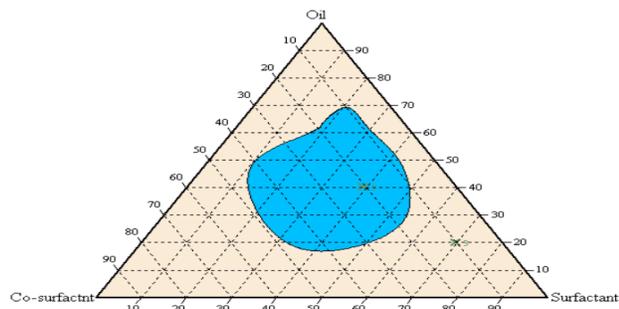


Fig. (6). Ternary phase diagram for Corn oil - Caproic acid - PEG600 with Smix in 3:1 ratio (Key: the filled region within the ternary phase diagram indicates nanoemulsification area where the transmittance is greater than 90).
 The mean globule size was decreased with increase in surfactant concentration. Hence the systems containing Corn oil - Caproic acid - PEG600 with 3:1 Smix ratio were selected for further studies due to their larger nanoemulsifying area, greater capacity for incorporation of oily phase with uniformity of dispersion and high transmittance values.

J. Effect of nebigivolol loading

Incorporation of nebigivolol (2.5 mg, 5 mg and 10 mg) led to a considerable decrease in transmittance values (figure 7, 8 and 9). This behaviour could be thought that undissolved drug in the compositions affected the clarity and thereby transmittance value to decrease with increased nebigivolol amount. Oil globules were observed on the surface after dispersion on standing for majority of the compositions containing high nebigivolol. The area of nano emulsification was considerably reduced with increase in nebigivolol loading in to the Corn oil - Caproic acid - PEG600 system with 3:1 Smix ratio hence for the stability reasons of the SNEDDS, system containing 10 mg of nebigivolol was chosen for formulation of nebigivolol SNEDDS and further studies.

CHEMIX School - TERNARY PLOT

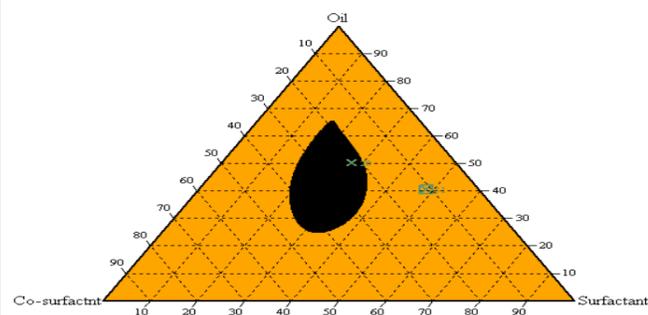


Fig. (7). Ternary phase diagram for 2.5 mg of nebigivolol loaded in Corn oil - Caproic acid - PEG600 system with Smix in 3:1 ratio (Key: the filled region within the ternary phase diagram indicates nanoemulsification area where the transmittance is greater than 90)

CHEMIX School - TERNARY PLOT

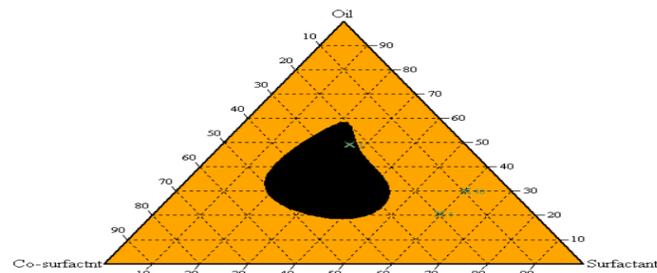


Fig. (8). Ternary phase diagram for 5 mg of nebigivolol loaded in Corn oil - Caproic acid - PEG600 system with Smix in 3:1 ratio (Key: the filled region within the ternary phase diagram indicates nanoemulsification area where the transmittance is greater than 90)

CHEMIX School - TERNARY PLOT

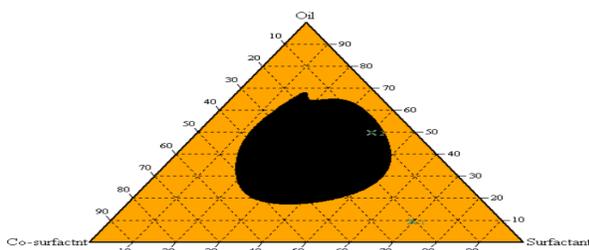


Fig. (9). Ternary phase diagram for 10 mg of nebigivol loaded in Corn oil - Caproic acid - PEG600 system with Smix in 3:1 ratio (Key: the filled region within the ternary phase diagram indicates nanoemulsification area where the transmittance is greater than 90).

K. Preparation and Evaluation of nebigivol SNEDDS

From the above results it was found that corn oil concentration in the range of 24-75% w/w, Caproic acid in the range of 15-60% w/w and PEG 600 in the range of 5-20% w/w in 3:1 of oil: Smix ratio with 10mg loaded nebigivol drug produced the SNEDDS having the transmittance greater than 90, with good stability.

L. Evaluations

1) Visual Observations

Visual observations indicated that at higher levels of surfactant, the spontaneity of the self-emulsification process was increased. This may be due to excess penetration of water into the bulk oil causing massive interfacial disruption and ejection of droplets into the bulk of aqueous phase [21]. When a co-surfactant, PEG 600 was added to the system, it further lowered the interfacial tension between the o/w interfaces and also influenced the interfacial film curvature (table 2).

2) Turbidity Measurement

Turbidity values (NTU) have been reported to be of use in SNEDDS characterization [22]. From these results it can be generalized that the formulations that have low turbidity (<20) gave a transmittance value of more than 90 indicating rapid and spontaneous emulsification within 1min, hence it gives a good correlation between transmittance and turbidity values (table 2).

TABLE 2: VISUAL OBSERVATION AND TURBIDITY MEASUREMENT VALUES

Formulation code	Visual Observation	Turbidity (NTU)
F1	A	16.13
F2	A	15.11
F3	A	17.78
F4	B	20.48
F5	B	15.34
F6	A	14.69
F7	A	16.81
F8	B	15.72
F9	A	14.80
F10	A	16.54
F11	A	19.66
F12	B	17.22
F13	B	18.45
F14	A	14.19

3) *Robustness to Dilution*

Nanoemulsions resulting from the dispersion of nebivolol SNEDDS (F1-F14) with distilled water, 0.1N HCl, pH 4.5 acetate buffer and pH 6.8 phosphate buffer were found to be robust to all dilutions and no separation or drug precipitation was observed even after 24 hours of storage.

4) *Percentage drug content and Entrapment efficiency*

The drug content of all formulations ranged between 95.65±0.62 to 99.23±1.14% with maximum value exhibited by F14 (table 3). The entrapment efficiency of all formulations varies between 94.32±0.92 to 98.51±1.02% with maximum value displayed by F14 (table 3).

TABLE 3: % DRUG CONTENT AND % ENTRAPMENT EFFICIENCY VALUES

Formulation code	%Drug content	% Entrapment Efficiency
F1	97.90±0.65	97.05±1.15
F2	98.42±1.15	97.91±0.51
F3	97.23±0.14	96.41±1.02
F4	95.65±0.62	94.32±0.92
F5	98.30±1.19	97.67±1.52
F6	99.02±1.49	98.25±1.77
F7	97.75±1.78	96.76±1.83
F8	98.04±1.15	97.31±1.29
F9	98.85±1.66	98.07±1.43
F10	97.84±1.45	96.98±1.95
F11	96.71±1.13	96.05±1.19
F12	97.42±1.62	96.54±1.25
F13	97.11±1.89	96.22±1.54
F14	99.23±0.14	98.51±1.02

Above parameters are communicated as Average ± Standard Deviation; (n=3)

M. In Vitro Dissolution Tests

Faster release rates were observed for nebivolol SNEDDS than the pure drug. Nebivolol SNEDDS F1-F14 released more than 60% of drug within 30min while Pure drug released only 30.25% of drug in 60mins. Formulation F14 exhibited highest drug release of 99.96% in 60min. The release of the drug from SNEDDS formulation was increased proportionally with increase in surfactant concentration and hence F14 exhibited high drug release (Figure 10).

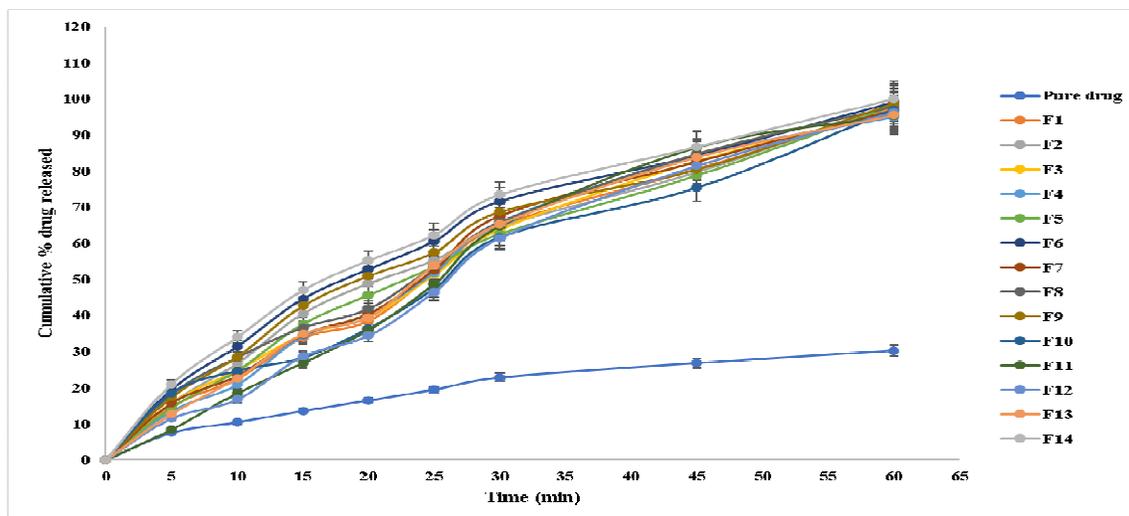


Fig 10: Comparative dissolution profile of nebigivol pure drug and nebigivol SNEDDS formulation (F1-F14)

N. Characterization of optimised formulation

1) FTIR Spectroscopy:

The characterization of pure drug nebigivol by FTIR studies was shown in fig.11. The spectrum is responsible for the presence of chemical functional groups at different frequencies. The pure Nebigivol spectrum showed the main characteristic bonds at 3201.94 cm^{-1} (O-H stretch), 3844.26 cm^{-1} (NH-bonding), 3726.60 cm^{-1} (OH-bonding), 1500 cm^{-1} (C-C stretch), 1433.16 cm^{-1} (C-H stretching), 1141.90 cm^{-1} (C-O-C stretching), 1257.63 cm^{-1} (C-O stretch), 1074.39 cm^{-1} (F-bonding), 815.92 cm^{-1} (C-X stretch), 1141.90 cm^{-1} (C-F stretch), 1618.33 cm^{-1} , a peak was found in the range of 1700–1750 cm^{-1} due to carboxylic acid moiety which indicates the presence of C=O stretching. The presence of prominent characteristic peaks confirming the purity of nebigivol as per the established standards. Frequencies of functional groups and unique absorption bands of pure drug remained intact in SNEDDS formulation (F14) indicate no significant interaction between drug and SNEDDS components (Figures 12, 13, 14 and 15).

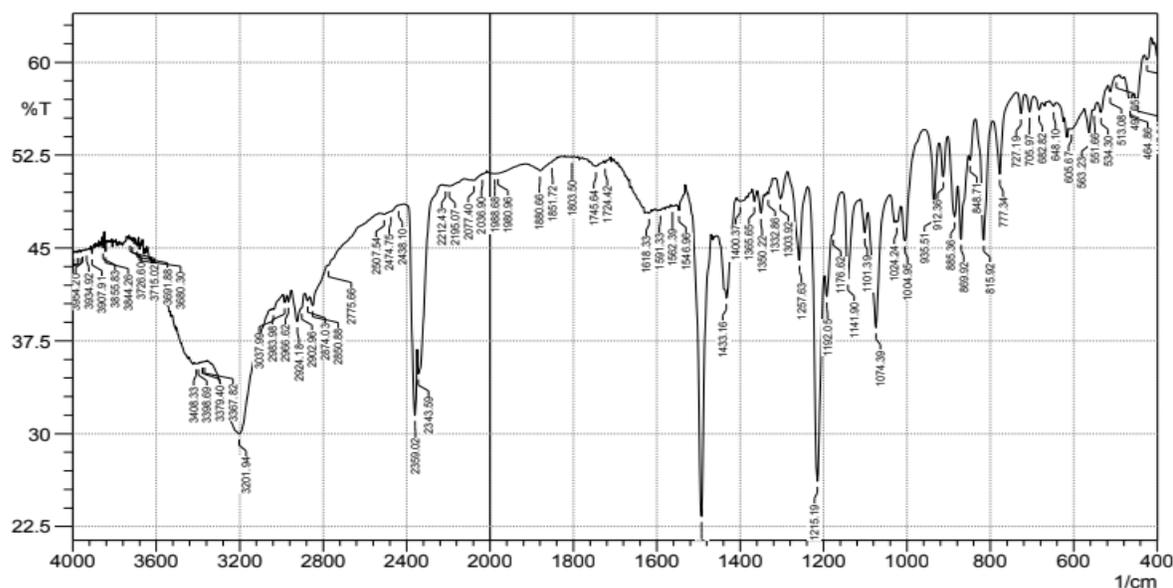


Fig 11: FTIR spectrum of pure drug nebigivol

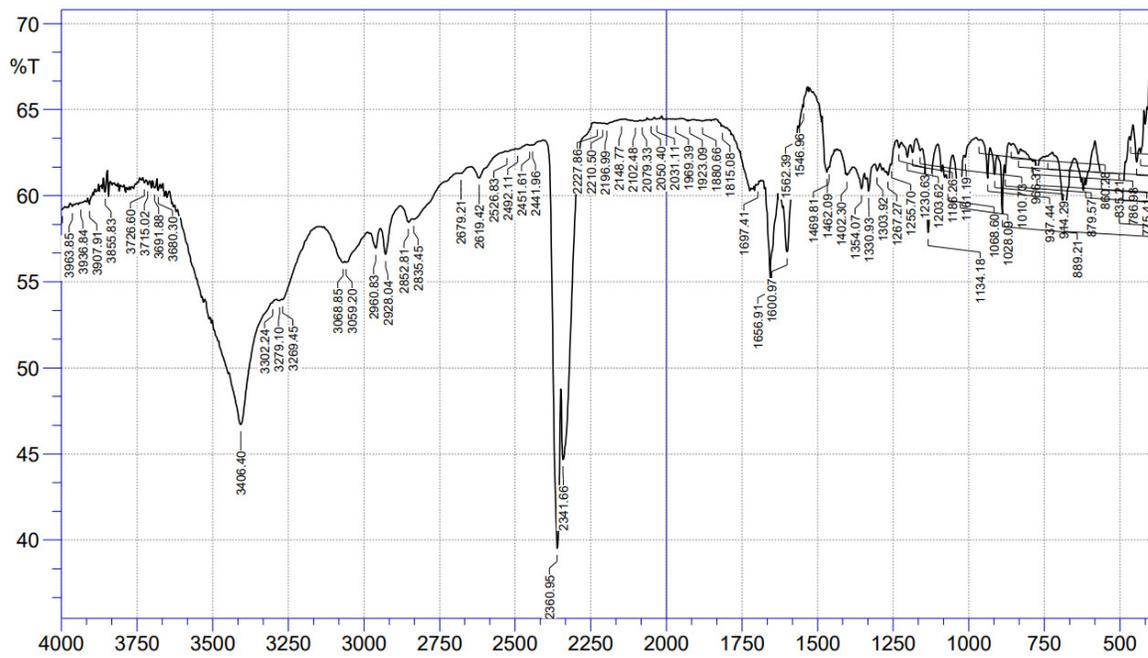


Fig 12: FTIR spectrum of Corn oil

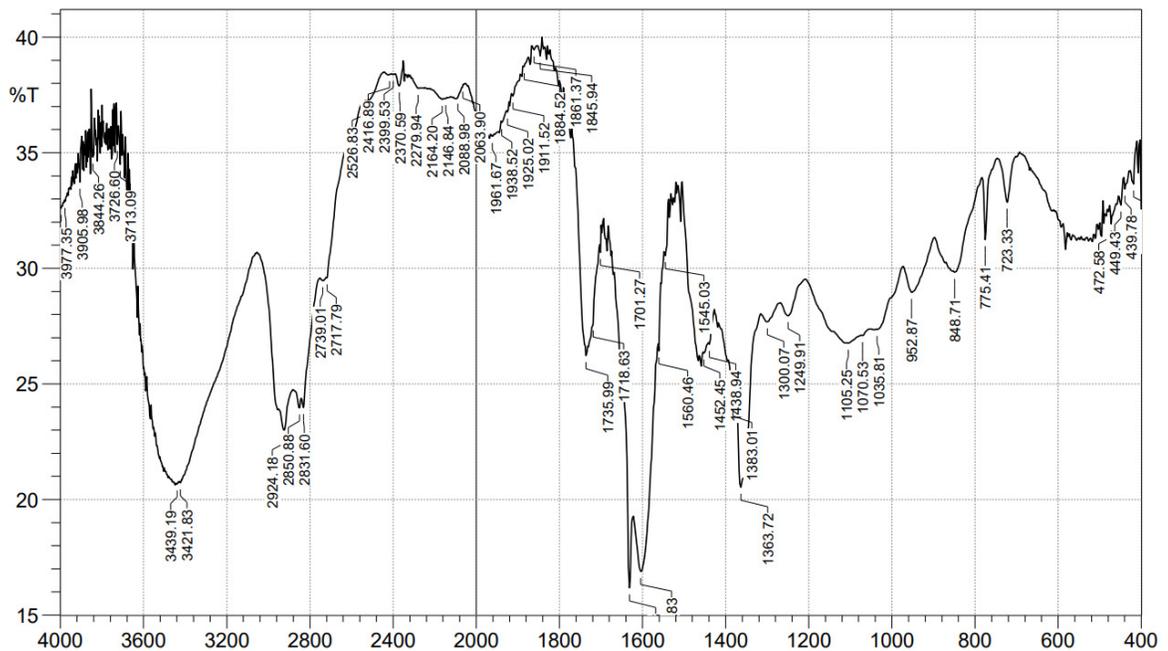


Fig 13: FTIR spectrum of Caproic acid

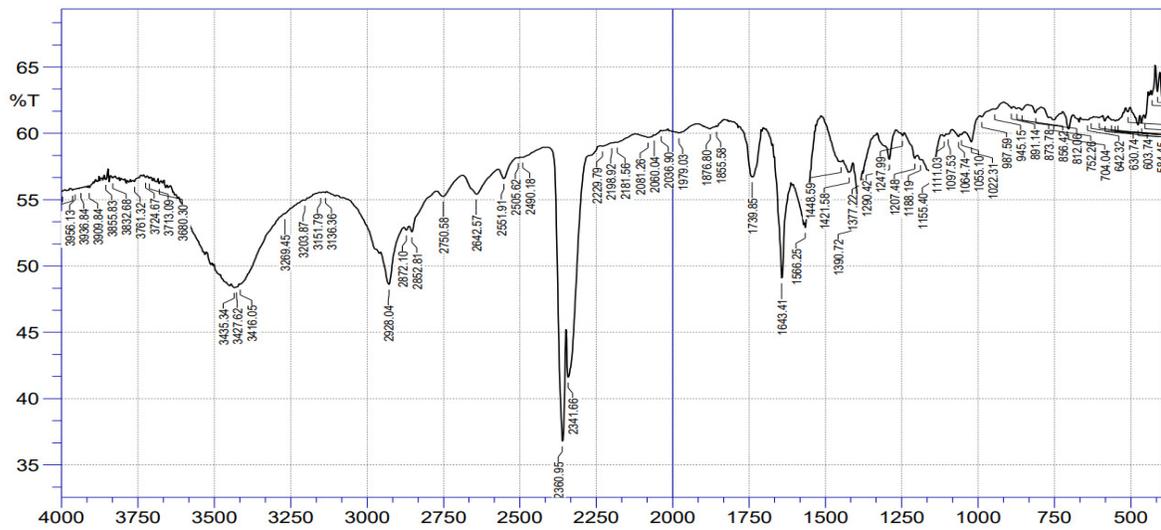


Fig 14: FTIR spectrum of PEG 600

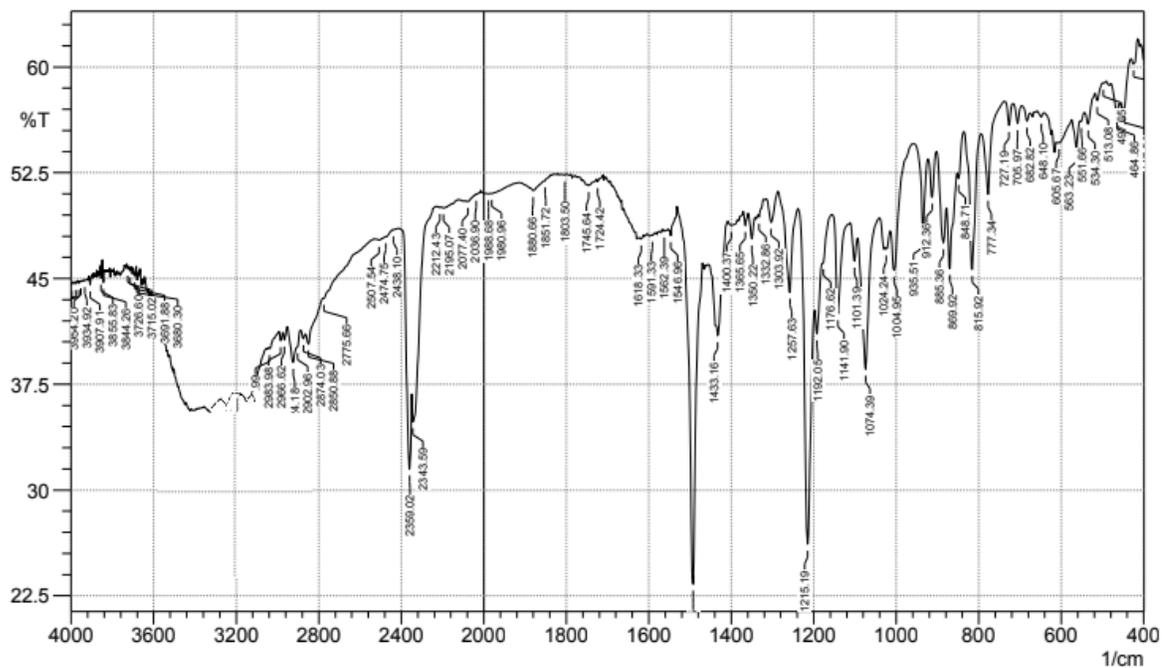


Fig 15: FTIR spectrum of optimised formulation of nebigolol SNEDDS (F14)

2. Globule size and zeta potential

The mean globule size of F14 was 165.2 nm indicating nanoparticle range that facilitates absorption. The zeta potential (mean) values of SNEDDS formulations were found to be in between -13.2 mV. The zeta potential value > 5 mV provide an excellent stability. [23] (Figures 16 and 17)

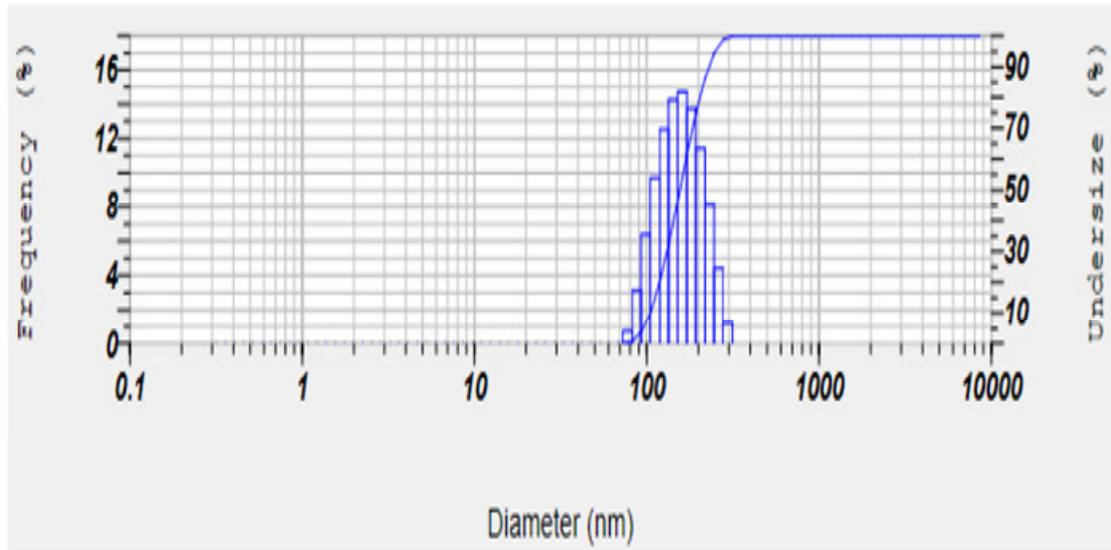


Fig 16: Particle size of optimised SNEDDS formulation of nebivolol (F14)

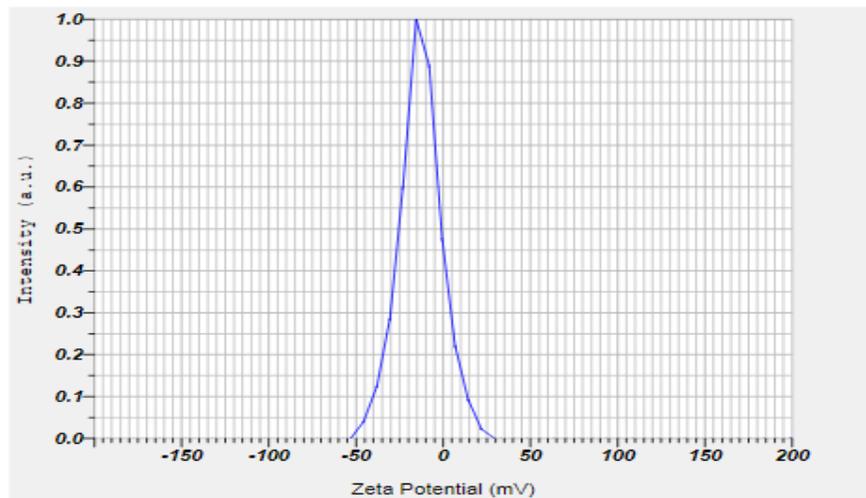


Fig 17: Zeta potential of optimised SNEDDS formulation of nebivolol (F14)

111

3. SEM studies

Figure 18A and 18B show scanning electron microscopic images of the nebivolol-optimized SNEDDS formulation F14. The formulation appeared to be spherical and smooth-surfaced, and globule size analysis confirmed these findings, with all droplets being less than 100 nanometres.

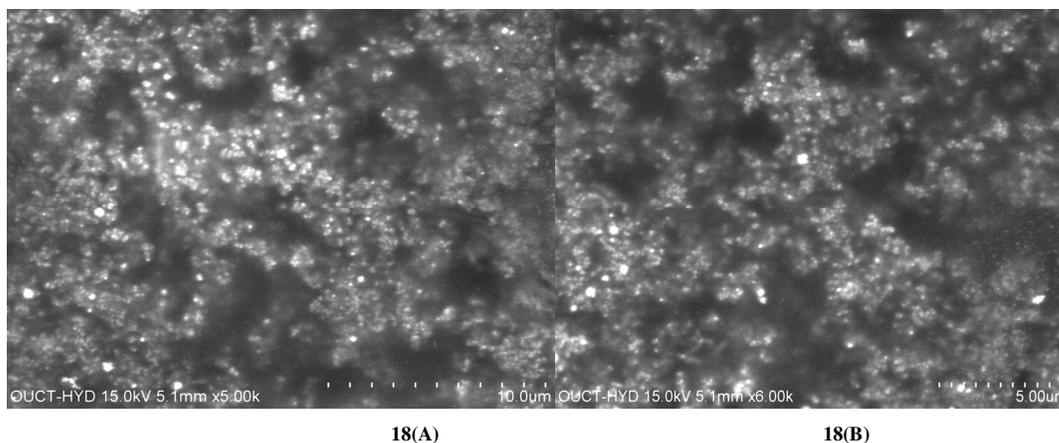


Fig 18: SEM images of optimised formulation of nebigolol SNEDDS F14 (A and B)

O. Forced Degradation and Accelerated Stability Studies of Optimized SNEDDS

Forced degradation and accelerated stability studies were conducted in order to know the stability of nebigolol in the gastric environment and in SNEDDS formulation. The degradation of nebigolol from optimised nebigolol SNEDDS formulation was significantly less when compared to the pure drug degradation. (Table 4)

TABLE 4: Percent Degradation of nebigolol from Pure Drug and Optimized nebigolol SNEDDS in Forced Degradation Study

Formulation code	Time (hr) / Diluting Solvent	% Drug Degraded (% , mean ± SD, n=3)				
		0 Hour	4 th Hour	6 th Hour	12 th Hour	24 th Hour
Pure drug	Methanol	0.02 ± 1.55	0.011 ± 0.24	0.01 ± 1.92	0.010 ± 1.27	0.10 ± 0.65
	Water	0.01 ± 1.25	0.01 ± 1.42	0.02 ± 1.65	0.04 ± 0.84	0.05 ± 1.59
	0.1 N HCl	0.07 ± 1.35	26.85 ± 1.22	28.33 ± 1.06	43.74 ± 1.35	62.31 ± 1.92
	pH 6.8 phosphate Buffer	0.01 ± 0.27	0.02 ± 1.12	0.12 ± 1.44	0.08 ± 1.07	0.07 ± 1.96
F14	Methanol	0.01 ± 1.85	0.12 ± 0.59	0.17 ± 0.62	0.02 ± 1.01	0.01 ± 1.01
	Water	0.02 ± 1.19	0.28 ± 0.84	0.01 ± 1.85	0.12 ± 1.43	0.07 ± 1.22
	0.1 N HCl	0.01 ± 1.75	0.05 ± 1.12	12.47 ± 1.07	28.12 ± 1.26	33.85 ± 2.07
	pH 6.8 phosphate Buffer	0.02 ± 1.22	0.05 ± 0.41	0.07 ± 1.74	0.04 ± 0.65	0.10 ± 1.35

Above parameters are communicated as Average \pm Standard Deviation; (n=3)

P. Accelerated Stability Studies

No visible physical changes were observed in all the formulations withdrawn from the humidity chambers. The samples were assayed for %entrapment efficiency, % drug content and in-vitro drug release and the results are shown in Table 5. No significant difference was observed after storage at accelerated conditions at $40\pm 2^\circ\text{C}/75\pm 5\% \text{RH}$ for a period of six months.

TABLE 5: STORAGE AT $40\pm 2^\circ\text{C}/75\pm 5\% \text{RH}$ FOR 6 MONTHS

Retest time for optimized formulation F14	% Drug content	% Entrapment efficiency	In-vitro drug release (%)
0 days	99.23 \pm 0.14	98.51 \pm 1.02	99.96 \pm 0.59
30 days	99.12 \pm 0.36	98.38 \pm 0.81	99.72 \pm 1.07
60 days	98.97 \pm 0.19	98.27 \pm 0.46	99.58 \pm 0.68
90 days	98.81 \pm 0.27	98.13 \pm 0.86	99.35 \pm 1.92

Above parameters are communicated as Average \pm Standard Deviation; (n=3)

IV. CONCLUSION

In our research, the potency of a self-nanoemulsifying drug delivery system was successfully investigated in order to provide an effective system for delivery of nebivolol. Based on the results of the solubility tests in various excipients, we selected the optimal oil, cosolvent, surfactant, and cosurfactant. Through the construction of a ternary phase diagram, the optimal formulation of nebivolol SNEDDS containing nebivolol was found to be Corn oil – caproic acid-PEG600. The particle size, zeta-potential and in vitro dissolution of the resultant emulsion after self-emulsification were determined. The average droplet size of the optimal formulation was 165.2 nm, and the average zeta-potential was -13.2 mV. SNEDDS- has a smaller particle size enhanced drug release of about 99.96% in 60 min which is about 3-fold higher than that of pure drug (32%). Based on forced degradation and stability studies, the SNEDDS can be considered as suitable formulation approach for improving the therapeutic efficiency of nebivolol.

REFERENCES

1. Date AA, Desai N, Dixit R, Nagarsenker M. Self-nanoemulsifying drug delivery systems: *formulation insights, applications and advances*. *Nanomedicine* (Lond) 2010; 5:1595–1616.
2. Date AA, Nagarsenker MS. Design and evaluation of self-nanoemulsifying drug delivery systems (SNEDDS) for cefpodoxime proxetil. *Int J Pharm*. 2007; 329:166–172.
3. Mohsin K, Long MA, Pouton CW. Design of lipid-based formulations for oral administration of poorly water-soluble drugs: precipitation of drug after dispersion of formulations in aqueous solution. *J Pharm Sci*. 2009; 98:3582–3595.
4. Raj, A. & Kumar, Y. (2018). Preparation and Evaluation of Solid Dispersion of Nebivolol Using Solvent Evaporation Method. *International Journal of Pharmaceutical Sciences and Drug Research*. 10. 10.25004/IJPSDR.2018.100418.
5. Yosra SRE, Magda AE, Ossama YA, Self-nanoemulsifying drug delivery systems of tamoxifen citrate: design and optimization. *Int J Pharm*, 2009, 380:133–41
6. Feng G, Haijun Z, Jing H, et al. Self-microemulsifying drug delivery system for improved oral bioavailability of dipyridamole: preparation and evaluation. *Arch Pharm Res*, 2011 34:1113–23
7. Czajkowska-Kośnik, A., Szekalska, M., Amelian, A., Szymańska, E., & Winnicka, K. Development and Evaluation of Liquid and Solid Self-Emulsifying Drug Delivery Systems for Atorvastatin. *Molecules*, 2015, 20(12), 21010–21022
8. Shiva Kumar Mantri, Shailaja Pashikanti and K. V. Ramana Murthy, “Development and Characterization of Self-Nanoemulsifying Drug Delivery Systems (SNEDDS) of Atorvastatin Calcium”, *Current Drug Delivery* 2012, 9: 182.

9. Sunny R. Shah, Rajesh H. Parikh, Jayant R. Chavda, et al. Self-Nanoemulsifying Drug Delivery System of Glimepiride: Design, Development, and Optimization. *PDA J Pharm Sci and Tech* 2013, 67 201-213
10. Kommuru, T.R.; Gurley, B.; Khan, M.A.; Reddy, I.K. Self-emulsifying drug delivery systems (SEDDS) of coenzyme Q10: formulation development and bioavailability assessment., *Int. J. Pharm.*, 2001, 212, 233-246.
11. Ping Z, Ying L, Nianping F, Jie X. Preparation and evaluation of self-microemulsifying drug delivery system of oridonin. *Int J Pharm* 2008, 355:269–76
12. Xiao, Y., Liu, Y., Yang, S., Zhang, B., Wang, T., Jiang, D., ... Zhang, N. Nebivolol and gadolinium co-loaded liposomes for drug delivery and MRI-guided HCC treatment. *Colloids and Surfaces B: Biointerfaces*, 2016, 141, 83–92.
13. Baratham, Srinivasa Rao & Ramanamma, Ch & Chowdary, K. Development of self-emulsifying drug delivery system of nebivolol for the improvement of solubility and dissolution, 2016, 5. 10.20959
14. Anusha, Ramineni. "Formulation and Evaluation of Nebivolol Tosylate Film Coated Tablets." (2012).
15. Reddy BS, Harish G, Md. Ul-Haq F: Formulation and *in-vitro* Characterisation of Solid - Self Nanoemulsifying Drug Delivery System (S-SNEDDS) of Rilpivirine. *Int J Pharm Sci Res* 2016; 7(7): 3117-29.
16. .Chopade, V.V.; Chaudhari, P.D. Development and evaluation of self-emulsifying drug delivery system for lornoxicam. *IJRDP* 2013, 2, 531–537
17. ICH Harmonized Tripartite guideline on "Stability Testing of New Drug Substances and Products Q1A (R2)", 6 February 2003
18. Reiss, H. Entropy-induced dispersion of bulk liquids. *J. Colloid Interface Sci.*, 1975, 53, 61-70.
19. Craig, D.Q.M.; Barker, S.A.; Banning, D.; Booth, S.W. An investigation into the mechanism of self-emulsification using particle size analysis and low frequency dielectric spectroscopy. *Int. J. Pharm.*, 1995, 114, 103-110.
20. Constantinides, P.P.; Scalart, J. Formulation and physical characterization of water-in-oil microemulsion containing long-versus medium-chain glycerides. *Int. J. Pharm.*, 1997, 158, 57-68.
21. Pouton, C.W. Formulation of self-emulsifying drug delivery systems. *Adv. Drug Deliv. Rev.*, 1997, 25, 47-58.
22. Nazzal, S.; Smalyukh, I.I.; Lavrentovich, O.D.; Khan, M.A. Preparation and *in vitro* characterization of a eutectic based semisolid self-nanoemulsified drug delivery system (SNEDDS) of ubiquinone: mechanism and progress of emulsion formation. *Int. J. Pharm.*, 2002, 235, 247-265
23. Kang, B.K.; Lee, J.S.; Chon, S.K.; Jeong, S.Y.; Yuk, S.H.; Khang, G.; Lee, H.B.; Cho, S.H. Development of self-micro emulsifying drug delivery systems (SMEDDS) for oral bioavailability enhancement of simvastatin in beagle dogs. *Int. J. Pharm.*, 2004, 274, 65-73.